Chemistry 3332
Name:
Signature: $\qquad$
ID\# $\qquad$

## PLEASE CIRCLE CLASS TIME!

## 10:00 AM

1:00 PM
1.


2.


3.

B. Facts: 19 points

1. Label the molecules below as aromatic (AR), antiaromatic (AA), or nonaromatic (NA). Please assume all are planar










2. Rank the following substituted benzene compounds in order of increasing reactivity by electrophilic aromatic substitution. (1=least reactive, 3=most reactive) (3 pts).





$\square$
3. Rank the following compounds in order of increasing reactivity with a nucleophile. (1=least reactive, 3=most reactive) (3pts)

$\square$

$\square$


4. Place the following compounds in order of increasing frequency of the $\mathrm{C}=\mathrm{O}$ stretch. (1=lowest frequency, 3=highest frequency) (3 pts)





$\square$
C. Reactions: Total $=36$ points, 4 points each Please provide the major product or the reagents in the answer box. Be sure your drawing indicates stereochemistry if applicable. Partial credit is awarded only when intermediate products in a multi-step reactio are shown below the reaction.
5. 



1. $\mathrm{H}_{2} \mathrm{SO}_{4}$ (fuming) $\xrightarrow{\text { 2. } \mathrm{Na} / \mathrm{NH}_{3} / \mathrm{CH}_{3} \mathrm{CH}_{2} \mathrm{OH}(\mathrm{xs})}$
2. 



$2 . \mathrm{Zn}(\mathrm{Hg}) / \mathrm{HCl}(\mathrm{aq})$

3.


4.


5.

6.

D. Mechanism (11 points)

Provide a clear mechanism to explain the formation of the product shown in the reaction below. Use curved arrows to indicate "electron flow". Remember to show only one step at a time. Show all intermediates and all formal charges.


## E. Synthesis (10 points)

Synthesize the molecule below using any of the following reagents: alcohols, alkanes,alkenes, and or alkynes of one carbon, cyclohexane, benzene, any inorganic reagents, any oxidizing or reducing agents, and any peroxyacids.


A compound with the formula $\mathbf{C}_{11} \mathbf{H}_{14} \mathbf{O}$ exhibits the IR and ${ }^{1} \mathrm{HMNR}$ shown below. Please identify this compound and draw the structure in the box provided below. ( 6 pts )





